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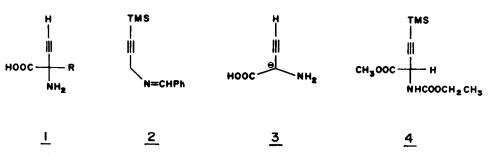
## TRIMETHYLSILYLACETYLENE-N-CARBOETHOXY GLYCINATE DIANION - A GENERAL SYNTHON FOR $\alpha$ -ACETYLENIC $\alpha$ -AMINO ACIDS.

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Mechanistic considerations suggest the use of  $\alpha$ -acetylenic amino acids <u>1</u> as potential enzyme-activated irreversible inhibitors of the corresponding  $\alpha$ -amino acid decarboxylases <sup>1</sup>. The recent syntheses of  $\alpha$ -acetylenic-3,4-dihydroxyphenyl alanine <sup>1,2</sup>, the sole known example of such an amino acid, involve the sequential alkylation and acylation of anions derived from the propargylamine synthon <u>2</u>. In view of the potential utility of this class of novel amino acids, an equivalent of the nucleophile <u>3</u> would be of obvious interest, as it would allow a variety of  $\alpha$ -acetylenic  $\alpha$ -amino acids to be prepared from a single precursor. An apparent approach would be to directly acylate the anion derived from <u>2</u>. This reaction, however, has been reported to afford an unidentified 2:1 adduct <sup>2</sup>. We now wish to report that the dianion prepared from the urethane <u>4</u> undergoes regioselective alkylation with a variety of electrophiles, and hence provides general synthetic access to the desired highly-funtionalized amino acids.



As shown below, the dianion which can be generated from <u>4</u> using excess lithium diisopropylamide/hexamethylphosphoramide (LDA/HMPA) undergoes alkylation with a representative series of alkyl halides. In no case were allenic products observed. The free  $\alpha$ -acetylenic  $\alpha$ -amino acids can be obtained from the alkylation products <u>5</u> by alkaline hydrolysis (2 M KOH, 12 hours reflux) or, more mildly, by first generating the isocyanate with SiHCl<sub>3</sub><sup>4</sup>, followed by alkaline hydrolysis (1 M KOH, 3 hours at 25°C).

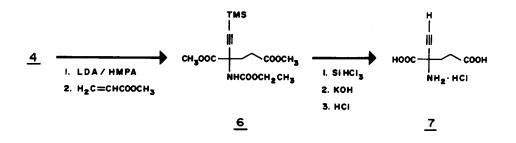
 $4 \xrightarrow{\text{TMS}} \\ H \xrightarrow{\text{H}} \\ 1. \ \text{LDA / HMPA} \\ 2. \ \text{RX} \\ \frac{5}{5} \\ \frac{1}{5} \\ \frac{1}$ 

	<u>R</u>	<u> </u>	Yield of 5
(a)	PhCH <sub>2</sub>	Br	75 %
(b)	H <sub>2</sub> C=CHCH <sub>2</sub>	Br	70 %
(c)	сн <sub>3</sub> (сн <sub>2</sub> ) <sub>3</sub>	I	60 %

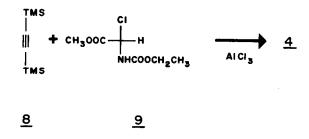
In a typical experiment <u>4</u> (1 mmol) in THF (5 ml) is added to LDA (3 mmol) in THF (10 ml) containing HMPA (1 ml) at -70°C. After 15 minutes at -70°C benzyl bromide (1 mmol) in THF (2 ml) is added. The mixture is maintained for 3 hours at -70°C, then quenched by the addition of acetic acid (2 mmol). The alkylation product <u>5a</u> (m.p. 97°C)<sup>5</sup> is obtained in 75 % yield after ion exchange chromatography and recrystallization.  $\alpha$ -acetylenic phenylalanine (<u>1</u> R = CH<sub>2</sub>Ph m.p. 184°C)<sup>5</sup> is then obtained by alkaline hydrolysis followed by ion exchange chromatography.

Although the alkylation by alkyl halides of the dianions derived from ethyl hippurate  $^6$  and from N-benzylbenzamide  $^7$  has been reported, in neither case was the conjugate addition to an  $\alpha$ , $\beta$ -unsaturated ester described. The

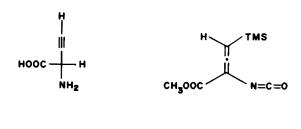
dianion from <u>4</u>, however, undergoes a regioselective 1,4-addition with methyl acrylate to afford the Michael adduct <u>6</u><sup>5</sup>, in 65% yield. Removal of the protecting groups then affords  $\alpha$ -acetylenic glutamic acid hydrochloride <u>7</u> (m.p. 160°C decomp.) <sup>5</sup>, as depicted below.



The glycinate derivative  $\underline{4}$  (m.p.  $49^{\circ}$ C) <sup>5</sup> is readily prepared in 65% yield by the amidoalkylation of bis-(trimethylsilyl)-acetylene ( $\underline{8}$ ) <sup>8</sup> with the 2chloro-N-carboethoxy glycinate ( $\underline{9}$ ), under Friedel-Crafts conditions (AlCl<sub>3</sub> (1 eq.) in dichloromethane for 12 hours at 25°C). The amidoalkylation of acetylenes usually leads to cyclic products which result from internal trapping of the intermediate vinyl cation <sup>10</sup>. This is avoided in this case, probably owing to the rapid departure of the trimethylsilyl group.



Attempts to prepare the parent amino acid of the series,  $\alpha$ -acetylenic glycine (<u>10</u>), by deprotection of <u>4</u> have proven abortive. Acid or base treatment results in unidentifiable products, while gentle deprotection of the urethane function using SiHCl<sub>3</sub><sup>4</sup> leads to the allene isocyanate <u>11</u>, which, although unstable, could be characterised spectroscopically.



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